

John A. Hyatt. Liquid and Supercritical Carbon Dioxide as Organic Solvents.

Page 5097. In a personal communication, G. Alexander (Monsanto Co., 800 N. Lindbergh Blvd., St. Louis, MO 63167) pointed out that, while valid solvent parameters such as $E_r(30)$ can be determined for supercritical CO_2 under a given set of conditions, it would be incorrect to consider any one $E_r(30)$ value as *uniquely* representing the supercritical fluid state of CO_2 as a whole. Alexander's caveat is supported by a recent paper in which Sigman, Lindley, and Leffler (*J. Am. Chem. Soc.* 1985, 107, 1471) report that the Kamlet-Taft π^* values for supercritical CO_2 vary with the density at which the measurements are made. The π^* and $E_r(30)$ values in my work were determined under conditions approximating constant density, and the reader should therefore understand that values for these parameters may diverge considerably at other densities.

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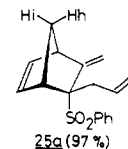
Srinivasan Nagarajan and Kenneth L. Rinehart, Jr.*. A Sigmatropic Rearrangement Involving Dimethyl Sulfoxide during an Oxidation of a Carbohydrate Derivative.

Page 380. In the supplementary material one of the cell parameters for compound **6** was omitted. It is $c = 11.221(2) \text{ \AA}$.

Kenji Hayakawa, Hitoshi Nishiyama, and Ken Kanematsu*. Reagent Design and Study of Allene as a Promising Class of Reagents (Synthons) for Cycloaddition. The Site Selective and Regioselective Diels-Alder Reactions of (Phenylsulfonyl)-propadiene and Alkylation of Adducts.

Page 513, right column, line 16. (*Jef*) should read (*Jed*).

Page 515, Table II, entry 3. The product is drawn incorrectly; it should be

**Gregory P. Mullen, Narendra P. Luthra, R. Bruce Dunlap, and Jerome D. Odom*.** Synthesis and Multinuclear Magnetic Resonance Study of Para-Substituted Phenyl Selenobenzoates.

Page 816.

We gratefully acknowledge the financial support of the National Institutes of Health (ESO-2836).

Mikiko Sodeoka and Masakatsu Shibasaki*. New Functions of (Arene)tricarbylchromium(0) Complexes as Hydrogenation Catalysts: Stereospecific Semihydrogenation of Alkynes and Highly Chemoselective Hydrogenation of α,β -Unsaturated Carbonyl Compounds.

Page 1148. Table I should read

Table I. Hydrogenation of Alkynes, α,β -Unsaturated Carbonyl Compounds and α,β -Unsaturated Imines

entry	starting material	no.	catalyst ^a	solvent	H ₂ press., kg/cm ²	temp, °C	time, h	product	no.	yield, %
1		1	MBZ-Cr(CO) ₃ ^b	acetone	70	120	23		2	92 ^c
2		1	NP-Cr(CO) ₃ ^d	THF	20	45	24		2	92 ^c
3		3	MBZ-Cr(CO) ₃	acetone	70	120	15			100 ^c
4		4	MBZ-Cr(CO) ₃	acetone	70	120	8			95 ^c
5		4	NP-Cr(CO) ₃	THF	50	45	8			87 ^c
6		5	NP-Cr(CO) ₃	THF	30	45	3		6	94 ^e
7		5	MBZ-Cr(CO) ₃	acetone	70	120	12		6	97 ^e
8		7	MBZ-Cr(CO) ₃	acetone	70	120	20	no reaction		
9		10	NP-Cr(CO) ₃	THF	70	45	26		11	96 ^e
10		12	NP-Cr(CO) ₃	THF	95	45	30		13	99 ^{c,f}
11		14	NP-Cr(CO) ₃	THF	80	120	12		15	11 ^{c,g}
12		16	MBZ-Cr(CO) ₃	acetone	70	120	24		17	93 ^e
13		18	NP-Cr(CO) ₃	THF	70	60	19		19	90 ^c

^aIn all the experiments, 0.2 molar equiv of the catalyst to the substrate was used. ^b(Methyl benzoate)tricarbylchromium. This commercially available catalyst requires high temperature and high H₂ pressure for the hydrogenation in general. ^cDetermined by gas chromatographic analysis relative to an internal hydrocarbon standard. ^dNaphthalenetetracarbylchromium. The hydrogenation can proceed under the milder conditions (45 °C) in the case of this catalyst (not commercially available). The required hydrogen pressure depends on substrates. ^eIsolated yield. ^fThe starting material (ca. 0.5%) was recovered. ^gThe starting material (69%) was recovered.